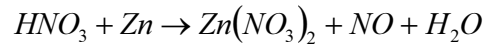
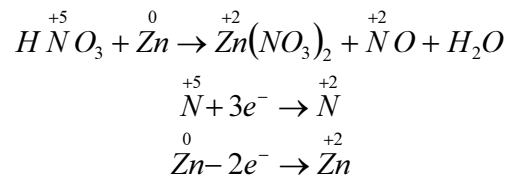


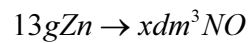
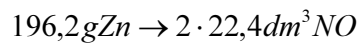
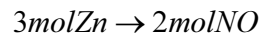
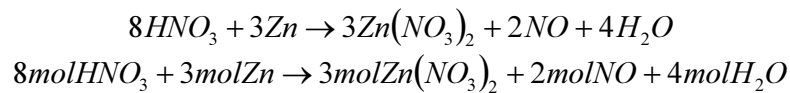
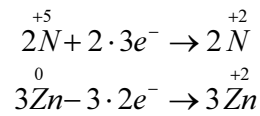
315. Koliko se litara NO izdvaja u reakciji $\text{HNO}_3 + \text{Zn} \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{NO} + \text{H}_2\text{O}$ dejstvom nitratne kiseline na 13 grama cinka? ($Zn=65,4$)



Ovo je oksido redukciona reakcija koju treba izjednačiti.



Kako broj primljenih elektrona mora da bude jednak broju otpuštenih, imamo:



$$x = 2,97\text{dm}^3\text{NO}$$