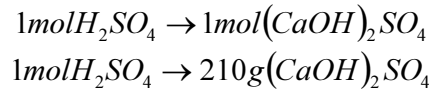
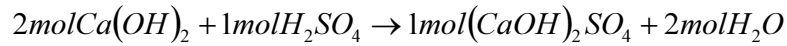
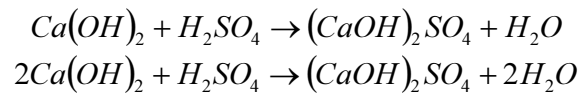
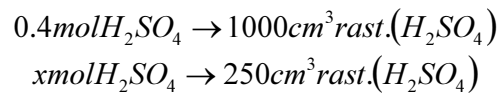


26. Koliko će se dobiti grama bazne soli u reakciji kalcijum hidroksida sa 250 mililitara rasvora sumporne kiseline koncentracije  $0,4\text{mol/dm}^3$  ? ( Ca-40, S-32 )



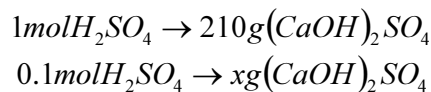
Na raspolaganju imamo rastvor  $\text{H}_2\text{SO}_4$  koncentracije  $0,4\text{mol/dm}^3$  što znači da:



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$$x = 0,1\text{molH}_2\text{SO}_4\text{u}250\text{cm}^3\text{rast.}(\text{H}_2\text{SO}_4)$$

Iz reakcije vidimo da :



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$$x = 21\text{g(CaOH)}_2\text{SO}_4$$