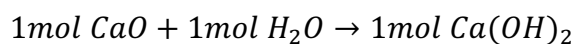
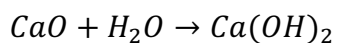
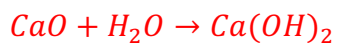
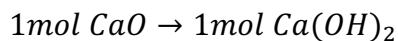


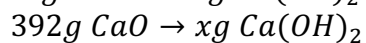
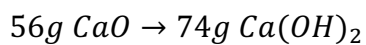
46. Koliko grama kalcijum hidroksida se dobije u reakciji 392g kalcijum oksida sa potrebnom količinom vode.



Iz ove reakcije vidimo da:



Pretvaramo molove CaO i Ca(OH)₂ u grame:



$$x = \frac{392\text{g CaO} \cdot 74\text{g Ca(OH)}_2}{56\text{g CaO}}$$

$$x = 518\text{g Ca(OH)}_2$$